













#### Concept of dust depletion

The depletion of an element X in the ISM is defined in terms of (a logarithm of) its reduction factor below the expected abundance relative to that of hydrogen if all of the atoms were in the gas phase,

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[Xgas/H] = \log\{N(X)/N(H)\} - \log(X/H)_{\odot}
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which is based on the assumption that solar abundances  $(X/H)_{\odot}$  are good reference values that truly reflect the underlying total abundances. In this formula, N(X) is the column density of element X and N(H) represents the column density of hydrogen in both atomic and molecular form, i.e.,  $N(H) + 2N(H_2)$ . The missing atoms of element X are presumed to be locked up in solids within dust grains or large molecules that are difficult to identify spectroscopically, with fractional amounts (again relative to H) given by

 $(X_{dust}/H) = (X/H)_{\odot}(1 - 10^{[X_{gas}/H]}).$ 

Jenkins 2009













# Cosmic Dust - "basic facts" 1/5 of the part of the Milky Way consisting of stars, planets and other "baryonic matter" is present as gas and dust. Of this 1/5 then 99% is gas and 1% is dust. Smoke particles consisting of C, O, Si, Mg, Al, Fe.

# General dust facts

Nucleation – requires supersaturation pressure

$$S = \frac{P_{mon}}{P_{sat}(T_d)}$$

- Change of dust mass:
  - Growth when smaller dust particle increase its size
  - Destruction when dust grain evaporates into gas phase
- No change of dust mass:
  - Shattering when larger grain is spilt into smaller dust grains
  - Coagulation smaller grains form larger grains
- Elements available for dust: C, O, Mg, Si, S and Fe

#### **Dust effects**

- Dust drives the mass loss of AGB stars.
- Dust is an important coolant during star formation, possibly essential for low mass star formation.
- Dust plays a crucial role for molecule formation in the ISM.
- Dust is an ingredient of planet formation.
- Dust determine "the weather" of BDs.
- Dust can be really irritating for certain observations!

#### History of the Universe

- Big Bang formed H (75%), He (25%), (Li, B, Be) No dust!
- Heavier elements are produced in stars.
- The abundance (by mass) today is H (74%), He (25%), all the rest ~ 1%









#### Extraterrestial samples

- Meteorites (from asteroids, Mars and the Moon).
- Lunar samples (returned by Apollo and Luna)
- Micrometeorites.
- Interplanetary dust particles (IDPs)
- Samples returned from comets (STARDUST)
- Samples from the Sun (Genesis)







- Chondrites 85.7% of falls age 4.55 billion years pristine samples of early solar system although in many cases their properties have been modified by thermal metamorphism or icy alteration.
- Subgroups: Enstatites contain the most refractory elements and are believed to have formed in the inner solar system. Ordinary chondrites, being the most common type containing both volatile and oxidized elements, are thought to have formed in the inner asteroid belt. Carbonaceous chondrites, which have the highest proportions of volatile elements and are the most oxidized, are thought to have originated in even greater solar distances.





Achondrites – 7.1% of falls - are also stony meteorites, but they are considered differentiated or reprocessed matter. They are formed by melting and recrystallization on or within meteorite parent bodies; as a result, achondrites have distinct textures and mineralogies indicative of igneous processes.

Subgroups: HED group thought to originate from asteroid Vesta. SNC group thought to originate from Mars. Aubrites which maybe originates from asteroid Nysa. Ureilites which maybe represents a c-type astroide.

#### Stony iron meteorites

Stony iron meteorites -1.5% of falls.

Subgroups: Pallasites composed of olivine enclosed in metal. Mesosiderites appear to be a surface regolith that has been stirred up and fused by repeated impacts.



#### Iron meteorites

Iron meteorites -5.7% of falls - are classified into thirteen major groups and consist primarily of iron-nickel alloys with minor amounts of carbon, sulfur, and phosphorus. These meteorites formed when molten metal segregated from less dense silicate material and cooled.





Iron and nickel form homogeneous alloys at temperatures below the melting point, these alloys are taenite. At temperatures below 900 to 600°C (depending on the Ni content), two alloys with different nickel content are stable: kamacite with lower Ni-content (5 to 15% Ni) and taenite with high Ni (up to 50%).















CI chondrites are very rare. Out of the ~ 1000 recorded observed meteorite falls from which material is preserved, only 5 CI chondrites are known. Among the 40,000 or so meteorites collected in Antarctica, only a few are CI chondrites.

The meteorites are very fragile and decompose easily, for example, if placed in water, CI chondrites immediately begin to disintegrate. Hence, CI chondrites that are found a long time after their fall are not useful for abundance studies, as chemical information is easily altered or lost.

Meteorite	Date of Fall	Country	Preserved Mass	
Alais	15 March 1806	France	6 kg	
Orgueil	14 May 1868	France	14 kg	
Tonk	22 January 1911	India	10 g	
Ivuna	16 December 1938	Tanzania	0.7 kg	
Revelstoke	31 March 1965	Canada	$\leq 1 g$	













Туре	Abundance (parts per million)	Size
Nanodiamond (C)	1400	2 nanometers
Silicon carbide (SiC)	14	0.1-20 micrometers
Graphite (C)	10	1-20 micrometers
Carbides of titanium, zirconium, molybdenum, ruthenium, and iron, and iron-nickel metal	Small grains inside presolar graphite	5-220 nanometers
ilicon nitride (Si <sub>3</sub> N <sub>4</sub> )	>0.002	About 1 micrometer
Corundum (Al <sub>2</sub> O <sub>3</sub> )	About 0.05	0.5-3 micrometers
Spinel (MgAl2O4)	<0.05	0.1-3 micrometers
libonite (CaAl <sub>12</sub> O <sub>19</sub> )	0.002	2 micrometers
Titanium dioxide (TiO <sub>2</sub> )	One grain	About 1 micrometer

## Nano-diamonds

Nano-diamonds precipitate as a cloudy white gel from acidic solution but they completely "dissolve" in basic solution.



Photo courtesy of Roy S. Lewis













## The δ-notation

Describes the deviation of an isotope ratio  $({}^{i}N/{}^{j}N)$  of a sample from the (terrestrial) standard ratio in per-mil:

 $\delta^{i}N(\infty) = [(^{i}N/^{j}N)_{sample}/(^{i}N/^{j}N)_{standard} - 1] \times 1000.$ 











Lodders & Amari 2005









Mineral	Abundance (~ ppm)	Typical Size Range (µm)	Stellar Sources	AGB stars are the	
Diamond	1500	0.002-0.003	SN ?	source o almost all SiC, Al <sub>2</sub> O <sub>3</sub> spinel and silicates.	
SiC	30	0.1-10	AGB (>90%) SN (1%) Novae (0.1%)		
Graphite	1	0.1-10	SN (80%) AGB (<20%) Novae (2%)		
Al <sub>2</sub> O <sub>3</sub>	0.2	0.1-5	RGB (>70%) AGB (20%) SN (<1%)		
Spinel MgAl <sub>2</sub> O <sub>4</sub>	50	0.1-5	RGB (>70%) AGB (20%)	e (2005)	
Silicates in IDPs	1000	0.1-1	RGB and AGB (>80%)	k Hoppe	
Silicates in Meteorites	100	0.1-1	RGB (>80%) AGB (10%)	e: Ott &	
Nitrides Si <sub>3</sub> N <sub>4</sub>	0.002	1	SN (100%)	Sour	

















Genesis was a NASA sample return probe which collected a sample of solar wind and returned it to Earth for analysis.

It was the first NASA sample return mission to return material since the Apollo Program, and the first to return material from beyond the orbit of the Moon.

Genesis was launched on August 8, 2001, and crash-landed in Utah on September 8, 2004, after a design flaw prevented the deployment of its drogue parachute.

The crash contaminated many of the sample collectors, and although most were damaged, many of the collectors were successfully recovered.[























#### Recommended reading

#### Books:

- C.R. Cowley, An Introduction to Cosmochemistry, Cambridge University Press, 1995
- H.Y. McSween Jr & G.R. Huss, *Cosmochemistry*, Cambridge University Press, 2010.

#### Articles:

- G.J. MacPherson & M.H. Thiemens, 2011, Cosmochemistry: Understanding the Solar System through analysis of extraterrestial materials, PNAS 108, pp. 19130-19134.
- E.K. Zinner, F. Moynier & R.M. Stroud, 2011, Laboratory technology and cosmochemistry, PNAS 108, pp. 19135-19141.
- A.M. Davis, 2011, Stardust in meteorites, PNAS 108, pp. 19142-19146.
- D.S. Burnett and Genesis Science Team, 2011, Solar composition from the Genesis Discovery Mission, PNAS 108, pp. 19147-19151.
- G.J. MacPherson & A. Boss, 2011, Cosmochemical evidence for astrophysical processes during the formation of our solar system, PNAS 108, pp. 19152-19158.
- T.J. McCoy, C.M. Corrigan & C.D.K. Herd, 2011, Combining meteorites and missions to explore Mars, PNAS 108, pp. 19159-19164.
- K.Righter & D.P. O'Brien, 2011, Terrestial planet formation, PNAS 108, 19165-19170.
- G.D. Cody, E. Heying, C.M.O. Alexander, L.R. Nittler, A.L.D. Kilcoyne, S.A. Sanford & R.M. Stroud, 2011, *Establishing a molecular relationship between chondritic and cometary organic* solids, PNAS 108, pp. 19171-19176.
- H.Y. McSween Jr, R.L. McNutt Jr & T.H. Prettyman, 2011, Spacecraft intrument technology and cosmochemistry, PNAS 108, 19177-19182.











23, 2011

24, 2011





16,000 km, July 17, 2011

July 18, 2011

This graphic shows the global distribution of craters that hit the giant asteroid Vesta, based on data from NASA's Dawn mission. The yellow circles indicate craters of (4 kilometers or wider, with the size of the circles indicating the size of the crater. The two huge impacts in the southern hemisphere appear as undulating lines in this projection.

## Meteorites - Asteroids

Are meteorites pieces of astroids?

(Antarctica).

If yes, why are the spectral characteristics so different? And how did the meteorites get from the Astroidebelt to Earth?

